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Stabilisation of Spent Wash by Polyhydroxybutyrate (PHB) Producing Microorganisms Isolated from Karad Region, Maharashtra

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ABSTRACT

In recent years, there has been a significant increase in the needs of the overgrowing population. Naturally, industrial belts increased worldwide to satisfy the variety and quantity of needs. While producing the products, a huge quantity of waste is generated and added continuously to the environment, causing hazardous effects on the flora and fauna. Spent wash is one of the most important types of industrial waste since it is the liquid waste that is left over after making alcohol and it pollutes the environment. Despite effluent quality standards, untreated or partially treated sewage frequently enters water courses or soils. It is always overloaded with organic and inorganic substances. The stabilization of the effluent by microorganisms is reflected in the form of COD reduction. The present study attempted to isolate the PHB-producing organisms and use them to stabilize the spent wash. The locally isolated, characterized, and identified 11 PHB-producing microorganisms from the Karad region of Maharashtra were used to stabilize spent wash. They include Candida orthopsilosis, Bacillus subtilis, Bacillus cereus, Bacillus megaterium, Klebsiella grimontii, Citrobacter fruendii, and members of Staphylococcus and Rhodococcus. All potential organisms were reported to stabilize the spent wash. The degree of stabilization was measured in terms of the percent reduction of COD and BOD. The % reduction was reported at 95.31% and 81.39% of COD and BOD, respectively, by Klebsiella grimontii, followed by 92.18% and 80.46% reductions of COD and BOD, respectively, by Staphylococcus. These organisms are promising in the treatment of spent wash.

INTRODUCTION

Interest in low-carbon and more sustainable energy production has grown in several nations worldwide due to worries about climate change, energy security, and the depletion of fossil fuels (Chooyok et al. 2013). Bioenergy is a carbon-neutral, renewable energy source that may be used to create any fuel. The need for sustainable and renewable biofuels, particularly bioethanol, is increasing globally, and ethanol distillery businesses are expanding quickly to meet this demand (David et al. 2015).

Molasses waste wash is the term for the high volume of strong wastewater produced by the ethanol distillery industry. The world's water quality and soil quality are seriously threatened by the ever-increasing output of molasses waste wash, which produces undesired residual liquid waste (Mohana et al. 2009). Distillery waste wash is a caramelized and recalcitrant waste with extremely high COD, BOD, SS, inorganic solids, color, and low pH. Melanoidins, which make up 2% of the total wasted wash by mass, are primarily responsible for the spent wash's deep, dark brown color (Naik et al. 2010). For melanoidin, including distillery effluent, color removal is difficult. Typically, 10-15 L of wastewater is produced for every liter of alcohol (Biradar 2003).

If distillery waste is disposed of improperly, it poses a major risk to the nearby land and water bodies because it blocks sunlight from penetrating the aquatic environment, depleting the dissolved oxygen concentration. This happened due to the melanoidin pigment, a dark brown pigment produced by the Maillard reaction between sugar and amino compounds (Pandey et al. 2003, Satyawali & Balkrishnan 2007, Khandegar & Saroha 2012). Investigations into how to decolorize it physically or chemically were unsuccessful. Utilizing microbial activity to decolorize the spent wash has received more attention recently. Environmentalists and academics face challenges in distillery waste wash's economical and environmentally responsible management. Distillery waste wash can be treated using various physical, chemical, and biological techniques, particularly for decolorization and COD removal. The method most frequently used to treat distillery waste wash is bio-methanation, followed by aerobic treatment (Yadav 2012). Chemical oxygen demand

(COD) and biological oxygen demand are reduced by aerobic treatment by up to 50-70%. (Shruti et al. 2013). Since biological treatment with bacterial cultures has a reduced operating cost, it might be a useful alternative to reduce COD in wastewater. In the current investigation, isolated PHB producers were utilized locally to reduce COD and BOD to stabilize the organic matter and make it easier to dispose of without endangering the local flora and fauna.

MATERIALS AND METHODS

Physiochemical Characterization of Distillery Spent Wash Sample

The effluent was analyzed in triplicate, before and after the addition of potential organisms, for its color, odor, pH, Chemical Oxygen Demand (COD) by Dichromate reflux method, and Biochemical Oxygen Demand (BOD) by Winkler's modified Alkali-Azide method (Greenberg et al. 1998).

Use of Potential Isolates for Stabilization of Effluent

Out of 75 locally isolated microorganisms with PHB production ability, 11 isolates with excellent PHB production potential (Table 1) were identified and used to test their abilities to stabilize the spent wash in terms of reduction in COD and BOD.

The COD Determination

The extent of organic matter degradation was determined mainly by the reduction in COD values. To determine the COD, 100 mL of mineral salt medium supplemented with 10% spent wash was inoculated with 1% inoculum containing 1×10^{8} CFU.mL⁻¹ potential PHB producers separately and incubated for 96 h at 30°C. After incubation, it was diluted

Table 1: Potential PHB-producing microorganisms.

| Sr. No. | Isolate No. | Isolate identified as | Accession Number |
|---------|----------------|--------------------------|---------------------|
| 1. | 1 | Candida orthopsilosis | OP393929 |
| 2. | 10 | Bacillus subtilis | OP415431 |
| 3. | 20 | Bacillus cereus | OP430573 |
| 4. | 24 | Member of Bacillus | - |
| 5. | 33 | Member of Rhodococcus | - |
| 6. | 37 | Klebsiella grimontii | OP458827 |
| 7. | 38 | Member of Bacillus | - |
| 8. | 41 | Citrobacter freundii | OP434597 |
| 9. | 45 | Member of Staphylococcus | - |
| 10. | 46 | Member of Staphylococcus | - |
| 11. | 72 | Bacillus megaterium | OP435363 |

to 1:1500, and samples were used to determine COD values by the dichromate reflux method (Hu & Grasso 2005). These values were then used to calculate the % reduction of COD, which directly revealed the extent of stabilization of effluent. The controls were without inoculum added.

The BOD Determination

The reduction in BOD values determined the extent of organic matter degradation. To determine the BOD, 100 mL of mineral salt medium supplemented with 10% spent wash was inoculated with 1 % inoculum containing 1×10^8 CFU.mL⁻¹ potential PHB producers separately and incubated for 96 h. After incubation, it was diluted, and samples were used to determine BOD values by modifying Winkler's alkali azide method (Sawyer & McCarty 1978, Greenberg et al. 1998) calculates the amount of oxygen absorbed by bacteria during organic matter breakdown. Five days at 20°C are used to measure the change in DO concentration. The BOD is determined by subtracting the starting DO from the final DO. These values were then used to calculate the % reduction of BOD, which directly revealed the extent of stabilization of effluent. The controls were without inoculums added.

RESULTS AND DISCUSSION

Physicochemical analysis of raw spent wash (Table 2) reflects its heavy organic load in terms of high values of COD and BOD. The color of the spent wash was observed as deep dark brown with an unpleasant odor and acidic pH.

The high values of COD and BOD in raw spent wash warn of the danger caused to the environment and make more attention to proper treatment to stabilize the organic load. This can be achieved effectively by the use of microbial treatment. In the present study, 11 potential PHB producers attempted to degrade the organic contents and reduce the COD and BOD.

Use of Potential Isolates for Stabilization of Spent Wash

The results of stabilization of spent wash in terms of COD value and % reduction are summarized in Table 3;

| Table 2: | : Physicoch | emical ana | lysis of | spent | wash | sample. |
|----------|-------------|------------|----------|-------|------|---------|
| | 2 | | 2 | | | 1 |

| Sr. No. | Parameter | Results |
|---------|-----------|---------------------------|
| 1. | Color | Deep dark brown |
| 2. | Odor | Unpleasant burn |
| 3. | pH | 5.5 |
| 4. | COD | 128000 mg.L ⁻¹ |
| 5. | BOD | 43000 mg.L ⁻¹ |



| Sr.No. Isolate | | COD value (after h) | | | | % COD re | % COD reduction (after h) | | | |
|----------------|-----|---------------------|--------|--------|--------|----------|---------------------------|-------|-------|-------|
| | No. | 24 | 48 | 72 | 96 | 24 | 48 | 72 | 96 | 120 |
| 1. | 1 | 1,20,000 | 90,000 | 48000 | 18,000 | 6.25 | 29.68 | 62.5 | 85.93 | 86.98 |
| 2. | 10 | 36,000 | 36,000 | 18,000 | 12,000 | 71.87 | 71.87 | 85.93 | 90.62 | 92.12 |
| 3. | 20 | 60,000 | 30,000 | 18,000 | 11,000 | 53.12 | 76.56 | 85.93 | 91.40 | 93.20 |
| 4. | 24 | 1,20,000 | 18,000 | 18,000 | 18,000 | 6.25 | 85.93 | 85.93 | 85.93 | 87.43 |
| 5. | 33 | 96,000 | 24,000 | 12,000 | 12,000 | 25 | 81.25 | 90.62 | 90.62 | 92.62 |
| 6. | 37 | 24,000 | 18,000 | 12,000 | 6,000 | 81.25 | 85.93 | 90.62 | 95.31 | 97.31 |
| 7. | 38 | 90,000 | 84,000 | 36,000 | 18,000 | 29.68 | 34.37 | 71.87 | 85.93 | 86.98 |
| 8. | 41 | 66,000 | 36,000 | 24,000 | 12,000 | 48.43 | 71.87 | 81.25 | 90.62 | 92.50 |
| 9. | 45 | 66,000 | 60,000 | 42,000 | 18,000 | 48.43 | 53.12 | 67.18 | 85.93 | 87.20 |
| 10. | 46 | 1,20,000 | 60,000 | 24,000 | 10,000 | 6.250 | 53.12 | 81.25 | 92.18 | 93.20 |
| 11. | 72 | 60,000 | 36,000 | 24,000 | 18,000 | 53.12 | 71.87 | 81.25 | 85.93 | 87.10 |

Table 3: COD value and % COD reduction of effluent after treatment with potential isolates. (Initial COD at 0 hr was 1,28,000 mg.L⁻¹).

From Table 3, it can be noticed that COD values decreased as the time of reaction increased. It is evident from the table that the % COD reduction for the isolate was maximum after 96 hrs. of incubation, while after 120 hrs. of incubation, the % COD reduction was meagerly increased up to 1-2%, and hence 96 hrs. of incubation was taken as values as after the period of reduction. Hence COD values due to remaining isolates were determined and considered only up to 96 hrs. The range of COD values at the end of 96 hrs was recorded between 18000 to 6000. When these values were compared with values of COD from the original sample, i.e., 128000, a tremendous decrease in COD was noticed. The range of % reduction in COD for all these 12 isolates was found to be 84.37% to 95.31%, which indicates the excellent efficiency of isolates for stabilization of spent wash in terms of % COD reduction. A maximum % reduction of COD, i.e., 95.31, was reported using isolate 37, identified as Klebsiella grimontii. As Mohana et al. (2009) noted, the considerable potential for pollution makes the release of untreated wasted wash effluent into the environment hazardous to the ecosystem. Because of this, the stability of spent wash that shows a decrease in COD is considerably more important.

Ramachandra (1993) noticed COD reduction by *Pseudomonas stutzeri*, *P. acidovorans*, and *Enterbacter* sp. *Alcaligenes eutrophus* growing in the spent wash. Kumar et al. (1997) reported 41% and 39% COD reduction by LA-1 and D-2 isolates under optimized conditions in eight days. Sharma et al. (2000) measured 60 % COD reduction by several lactic acid bacteria from anaerobically digested molasses spent wash (ADMSW). Sudarshan et al. (2012) utilized *Azotobacter vinelandii* ATCC 12837 and ATCC 13705 to reduce COD levels (1100 mg.L⁻¹) of DSW to 200 mg.L⁻¹ after fermentation was completed. Thus they successfully reduced the COD value to the permitted level.

In the present study, we have also achieved these two purposes of obtaining higher PHB quantity, and maximum COD reduction in distillery spent wash. Thus present results are on the same track of research giving suitable solutions for disposing of spent wash after proper treatment and sustaining the environment. The results of the stabilization of spent wash in terms of BOD value and % reduction are summarized in Table 4.

From Table 4, it is observed that values of BOD were recorded between 14200 to 8000. When these values were compared with values of BOD from the original sample, i.e., 43000, a large extent decrease in BOD was noticed. The range of % reduction in BOD for all these 11 isolates was found to be 66.98% to 81.39%, which indicated the excellent efficiency of isolates for stabilization of spent wash in terms of % BOD reduction. The maximum % reduction of BOD, i.e. 81.39 was reported using isolate number 37, identified as *Klebsiella grimontii* (Table 5).

Table 4: BOD value and % BOD reduction of effluent after treatment with potential isolates BOD value at 0 hrs was 43000 mg.L⁻¹.

| Sr.No. | Isolate Number | BOD Value mg.L ⁻¹ after 5 days | % Reduction of BOD |
|--------|-------------------|--|--------------------|
| 1. | 1 | 11400 | 73.48 |
| 2. | 10 | 9200 | 78.60 |
| 3. | 20 | 14200 | 66.98 |
| 4. | 24 | 10200 | 76.28 |
| 5. | 33 | 9200 | 78.60 |
| 6. | 37 | 8000 | 81.39 |
| 7. | 38 | 11400 | 73.48 |
| 8. | 41 | 8800 | 79.53 |
| 9. | 45 | 13800 | 67.90 |
| 10. | 46 | 8400 | 80.46 |
| 11. | 72 | 12800 | 70.23 |

Table 5: Physicochemical analysis of spent wash treated with Klebsiella grimontii.

| Sr. No. | Parameter | Results |
|---------|---------------|-----------------|
| 1. | Color | Faint brown |
| 2 | Odor | Unpleasant burn |
| 3. | pH | 5.7 |
| 4. | COD reduction | 95.31% |
| 5. | BOD reduction | 81.39% |

The pioneering research on bacterial wasted wash decolorization was carried out by (Kumar et al. 1997). They found that under ideal conditions, two aerobic bacterial isolates, LA-1 and D-2, achieved the highest levels of decolorization (36.5% and 32.5%) and COD reduction (41% and 39%) in just eight days. Ramachandra (1993) named four bacterial species as *Pseudomonas stutzeri*, *P*. acidovorans, Enterbacter sp., and Alcaligenes eutrophus, which were primarily found thriving in the wasted wash. They saw a significant reduction in BOD and COD and a decolorization of up to 60%. Sharma et al. (2000) isolated numerous lactic acid bacteria (ADMSW) from anaerobically digested molasses wasted wash.

Some isolates lowered COD by 60% and decolored ADMSW (25% v/v) by 70%. The viability of utilizing *Pseudomonas putida* for decolorizing used wash was discussed by Ghosh et al. (2003). The organism exhibited the capacity to extracellularly transform glucose into gluconic acid and hydrogen peroxide using glucose oxidase. The hydrogen peroxide created by this enzymatic oxidation removed the color from the used wash by oxidizing melanoidins. According to Sirianuntapiboon et al. (1988), Mycelia sterile D90 was the most effective strain, decolorizing molasses pigment by nearly 93% after eight days. When nutrients such as glucose (2.5%), NaNO₃ (0.2%), KH₂PO₄ (0.1%), and MgSO₄.7H₂O (0.05%) were added to molasses wastewater, this strain also induced a BOD drop of nearly 80%. However, when these nutrients were absent, the decolorization yield was only 17.5%.

Our present results of COD reduction are on the same line of research as that of Kumar et al. (1997), Ramachandra (1993), Sharma et al. (2000) and Ghosh et al. (2003). On the other hand, the results of BOD reduction are similar to the results of Sirianuntapiboon et al. (1988).

A high % reduction in COD and BOD reflects these isolates' abilities to degrade the effluent's organic content. Thus converting the original hazardous effluent, such as a form of effluent that is safe to discharge into public water bodies by a suitable division nearby Karad city and safeguarding flora and fauna of the ecosystem.

CONCLUSIONS

The isolated 11 PHB producers efficiently degrade the organic content and stabilize the spent wash. The Klebsiella grimontii has the potential to stabilize the spent wash to a maximum degree by achieving a 95.31% reduction in COD and an 81.39% reduction in BOD. Thus biological treatment using Klebsiella grimontii culture may be a good alternative to stabilize the spent wash to a good extent.

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